

## Math 1310: CLS **SOLUTIONS** Tutorial: Modeling with Differential Equations

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1. Two strains of bacteria – *Calcul S. Coli*, denoted  $C$ , and *Lifus S. Sciencus*, denoted  $L$  – are dissolved in a solution of Sprite, denoted  $S$ . Below, on this page, is a verbal description of the interactions among the two types of bacteria and the Sprite.

On the *next* page, you will find:

- (a) Space at the top of the page for writing down differential (rate) equations for  $C'$ ,  $L'$ , and  $S'$ ; and
- (b) A list, at the bottom of the page, of terms that might appear in these differential equations.

Your task is to *build* (and write down) the differential equations in part (a) out of the terms listed in part (b). Each term may be used *at most* once; some won't be used at all.

Here is the description of the interactions:

- $C$  grows logistically, with carrying capacity proportional to the amount of  $L$  present.
- Sprite consumes  $C$  at a rate proportional to the amount of  $C$  present.
- A pair of  $C$  bacteria can spontaneously join to form an  $L$  bacterium; the overall rate at which this occurs is proportional to the number of possible  $C$ -to- $C$  interactions;
- $L$  grows at a rate that is, in the absence of other factors, proportional to the amount of  $L$  present, but that is inhibited by  $S$ : the more  $S$  there is present, the more slowly  $L$  grows;
- An  $L$  bacterium can spontaneously split into two  $C$  bacteria; the overall rate at which this occurs is proportional to the amount of  $L$  present;
- $S$  grows at a rate proportional to the amount of  $C$  present times the amount of  $L$  present;
- Each *individual*  $C$  bacterium consumes  $S$  at a rate proportional to the amount of  $S$  present;
- $L$  consumes Sprite at a rate proportional to the amount of  $L$  present.

SOME HINTS:

- You should need exactly *ten* of the terms from part (b) to construct your differential equations in part (a).
- “The number of possible  $C$ -to- $C$  interactions” is proportional to  $C^2$ , since every  $C$  bacterium can interact with every other  $C$  bacterium.
- Since a pair of  $C$  bacteria can join to form an  $L$  bacterium, the rate at which  $C$  bacteria are lost to this process must be twice the rate at which  $L$  bacteria are produced by it.

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- Similarly, since an  $L$  bacterium can split to form a pair of  $C$  bacteria, the rate at which  $C$  bacteria are produced by this process must be twice the rate at which  $L$  bacteria are lost to it.

(a)

$$C' = +aC \left(1 - \frac{C}{bL}\right) - wC - 2\ell C^2 + 2tL$$

$$L' = +\frac{dL}{1+eS} + \ell C^2 - tL$$

$$S' = +gCL - vCS - hL$$

- (b) (In the terms below, all lower case letters are positive parameters, and all upper case letters are variables.) Not all terms will be used.

$$+aC \left(1 - \frac{CL}{b}\right) ; \quad -eS ; \quad -wC ; \quad +tL ; \quad +\ell C^2 ;$$

$$+2tL ; \quad -tL ; \quad -2\ell C^2 ; \quad +gCL ; \quad +aC \left(1 - \frac{C}{bL}\right) ;$$

$$-hL ; \quad -vCS ; \quad +\frac{dL}{1+eS} ; \quad +d(L - eS)$$

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2. Chocolate (C), vanilLa (L), and Strawberry (S) ice cream interact in a sundae, according to the differential equations below. By filling in the blanks below, describe their interactions in ways that are consistent with the terms in these differential equations.

$$C' = aCL - \frac{c}{1 + dS}$$

$$L' = kL\left(1 - \frac{L}{qS}\right) - fL$$

$$S' = nCS + 3fL$$

Chocolate ice cream grows at a rate proportional to the product of the amount of Chocolate and vanilLa ice cream present. Chocolate ice cream melts, but the melting is inhibited by Strawberry ice cream: the more Strawberry ice cream, the more slowly Chocolate ice cream melts.

vanilLa ice cream grows logistically, with carrying capacity proportional to the amount of Strawberry ice cream present. Also, a molecule of vanilLa ice cream can spontaneously melt into three molecules of Strawberry ice cream; this happens at a rate proportional to the amount of vanilLa ice cream present.

Strawberry ice cream grows at a rate proportional to the product of the amounts of Chocolate and Strawberry ice cream present. Strawberry ice cream doesn't melt. (And, again, three molecules of Strawberry ice cream can be spontaneously generated from one molecule of vanilLa ice cream.)